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1) :1-14 The

Parable of the

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of the Marriage
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Life?~~

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~~Memorizing
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Wedding Come,

Lord Jesus! |

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179-183 Book 2

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You are at
chapter 22!!

Let's goooo!!!!

First of all, a hydrocarbon is the simplest organic compound and it only contains carbon and hydrogen.

For this, we can

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use our background knowledge to know that since carbon has four valence electrons, it can form four covalent bonds. Also, since hydrogen only has one, it can only make one bond.

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Chemistry by
Anna

atom. Therefore,
the simplest
hydrocarbon
molecule
consists of a
carbon atom
bonded to four
hydrogen atoms,

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CH₄. This

substance is called methane and is an excellent fuel and the main component of natural gas. 698

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Figure 22-1

Carbon is a nonmetal element located in group

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4A of the periodic table.

Elements in this

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will compare the

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Hydrocarbons and

properties of

alkanes,

alkenes, and

alkynes. You

will recognize

and compare the

properties of

structural

isomers and

stereoisomers.

You will

describe how use-

ful hydrocarbons

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are obtained from natural sources. Chapter 22 Hydrocarbons -

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position of

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double bond at

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each carbon of

the double bond

cis-2-butene

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different atoms

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carbon atom.

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a controlled
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hydrocarbons are
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molecules.

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Chapter 22

Substituted Hydrocarbons

Hydrocarbons and
Their Reactions
Section 22.12.1

Figure 22.1 All
of these items
contain at least
one of the
functional
groups that you
will study in
this chapter.

For example, the
fruit and

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Chapter 22

flowers have sweet-smelling aromas that are due to ester molecules.

Objectives

Define

functional

group, and give examples.

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Hydrocarbons and

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chapter 22

hydrocarbons,
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Petroleum.

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Hydrocarbons
Complex mixture
of carbon
compounds
produced by heat
and pressure
acting on the
remains of once-
living organisms
buried deep
beneath Earth's
surface.
Vitalism.

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This unit looks at aliphatic and aromatic hydrocarbons, including alkyl substituted hydrocarbons and structural and geometric isomers. There is also a brief

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Introduction to
functional
groups. Chapter
22 - Hydrocarbon
Compounds

Chapters 22 & 23
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Compounds 241

position of

their atoms a

double bond at

least one

substituent on

each carbon of

the double bond

cis-2-butene

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superimposed

true It is a

carbon with four

different atoms

or groups

attached. They

contain an

asymmetric

carbon atom.

trans-2-butene

CH₂ CH₃

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Hydrocarbons;

22.2 Unsaturated

Hydrocarbons;

22.3 Functional

Groups; 22.4

Alcohols and

Ethers; 22.5

Amines and

Alkaloids; 22.6

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Polymers;

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Biomolecules Are

Molecules

Produced and

Used by

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Carbohydrates

Give Structure

and Energy

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unquestionably
simple means to

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Hydrocarbons

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ChemFacts. • The primary source of hydrocarbons is petroleum. • About 75 million barrels of petroleum are pumped out of the Earth each day. •

Hydrocarbons are used as fuels

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Hydrocarbons are the raw materials for products such as plastics, synthetic fibers, solvents, and industrial chemicals.

Chapter 21:

Hydrocarbons

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Hydrocarbons 700

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Chapter 22 Hydrocarbons

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Figure 22-4 This taxi in Bangkok, Thailand is one of many cars, trucks, and buses that have been modified to burn propane.

Burning propane benefits the environment because it

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produces less
air pollution
than does
gasoline. In
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SECTION 22.2

UNSATURATED

HYDROCARBONS

(pages 702–703)

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explains the
difference

between

unsaturated and

saturated Page

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expert in the field, the book provides an engineer with data they can trust. Spanning gases, liquids, and solids, all critical properties (including viscosity, thermal conductivity,

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and diffusion coefficient) are covered. From C1 to C100 organics and Ac to Zr inorganics, the data in this handbook is a perfect quick reference for field, lab or classroom usage. By collecting a large - but

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relevant – amount of information in one source, the handbook enables engineers to spend more time developing new designs and processes, and less time collecting vital properties data. This is not a

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treatise, but an aid to the practicing engineer in the field, on day-to-day operations and long range projects.

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trust in design,
research,
development and
manufacturing A
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critical
temperature
dependent
properties for a
wide range of
hydrocarbons,
including C1 to
C100 organics
and Ac to Zr
inorganics

Polycyclic

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Hydrocarbons are

of interest in
many fields of
science:

theoretical

chemistry,

physical

chemistry,

organic

chemistry,

dyestuff

chemistry and

biology. With

regards to the

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latter, I am indebted to Dr. Regina Schoental of the Medical Research Council for the review in this present work of carcinogenesis by polycyclic hydrocarbons. This book is designed to present the

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facts in a simple and clear order and to derive empirical rules from them, but it does not present a comprehensive theory about polycyclic hydrocarbons. An attempt is made instead to extend classical

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symbolism into
modern
structural
chemistry. Thus
extensive use is
made of
Robinson's
aromatic sextet,
which is applied
in an
uncompromising
and strict way.
This quasi-
classical

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attempt is encouraged further by such completely unexpected discoveries as those of Dewar benzene and of the electronic asymmetry of formally symmetric hydrocarbons. How difficult it

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is to break away
from any
established way
of thinking has
been admirably
expressed by
Kekule

("Organische
Chemie", 1861,
Part 1, page 4,
translated from
German): "All
our ideas are
based, to an

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Hydrocarbons
extent much

greater than we ordinarily believe, on those of our predecessors.

Our accumulated experience, the notions of which our training has accustomed us to, of whatever kind they have been, influence

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the course of our thoughts far more than we are willing to admit; only too frequently the following of our regularly used, well trodden way of thinking leads to us overlook the simplest of correlations.

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the subject and
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approaches. This
book attempts to
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needs of
graduate and
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students and
other
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food, soil, water, and air pollution. The aim of this book is to explain and clarify important studies, and compare and develop the new and groundbreaking measurement techniques.

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comprehensive
examples.

The Oxidation of
Hydrocarbons in
the Liquid Phase
investigates the
kinetics and
mechanism of the
oxidation of
hydrocarbons in
the liquid

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phase. Topics covered include the chain reactions taking place in the liquid-phase oxidation of hydrocarbons; the action of inhibitors on degenerately branched chain reactions; and the mechanism of

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Inhibitor action
during
oxidations with
molecular
oxygen. The
liquid-phase
oxidation of
some aryl-
aliphatic
hydrocarbons is
also discussed.
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comprised of 35
chapters and

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liquid-phase hydrocarbons

oxidation

processes is

then considered,

along with

several

elementary

stages of the

liquid-phase

oxidation of

hydrocarbons.

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chapters explore

the relationship

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between the length of unbranched chains and the concentration of oxidation products accumulating in the reaction mixture on the basis of the theory of degenerate chain branching; the

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effect of
various

inhibitors on
the

autocatalytic

oxidation of

cycloparaffins;

the role of

metals during

the liquid-phase

oxidation of

fuels; and the

oxidation of

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with oxygen. The final chapter examines the behavior of lubricating oils in internal combustion engines, paying particular attention to the influence of temperature and oxygen on the

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oxidation of the
oil both in bulk
and in thin
films. This
monograph will
appeal to
chemists and
engineers and
those concerned
with the
establishment of
a detailed
mechanism of
liquid-phase

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oxidation
reactions.

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regulatory problems, and environmental concerns

constantly

challenge

chemists to

rethink and

redesign the

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the field,
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current
research, and
future
applications.
Hydrocarbon
Chemistry begins
by discussing
the general
aspects of
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the separation

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isomerization,
and alkylation

Derivation
reactions to
form carbon-
heteroatom bonds

Hydrocarbon
oxidations

Metathesis

Oligomerization
and

polymerization

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of hydrocarbons

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have been

updated by

adding sections

on recent

developments to

review new

advances and

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'exploration' geophysics on all scales. It forms an introduction to

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geophysics
suitable for
those who do not
necessarily
intend to become
professional
geophysicists,
including
geologists,
civil engineers,
environmental
scientists, and
field
archaeologists.

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The book is organised into two parts: Part 1 describes the geophysical methods, while Part 2 illustrates their use in a number of extended case histories.

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principles are introduced at an elementary level, and then developed as necessary.

Student questions and exercises are included at the end of each chapter. The book is aimed primarily at

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